

# Lab Activities for Lab 1

Your name: \_\_\_\_\_ Lab Partners \_\_\_\_\_

For calculation practice with scientific notation, see the handout sheet on scientific notation.

1. Measure tables (height, width, and length). Aim for 1 mm accuracy, Express in meters, centimeters, and millimeters.

\_\_\_\_\_ m x \_\_\_\_\_ m x \_\_\_\_\_ m  
 \_\_\_\_\_ cm x \_\_\_\_\_ cm x \_\_\_\_\_ cm  
 \_\_\_\_\_ mm x \_\_\_\_\_ mm x \_\_\_\_\_ mm

2. Measure thickness of this paper with micrometer. Aim for 20  $\mu\text{m}$  accuracy (smallest division on the micrometer). Express in m, cm, mm and  $\mu\text{m}$  using decimals and also scientific notation.

decimal: \_\_\_\_\_ m = \_\_\_\_\_ cm = \_\_\_\_\_ mm = \_\_\_\_\_  $\mu\text{m}$

scientific notation: \_\_\_\_\_ m

3. Measure a strand of hair (diameter). Aim for 20  $\mu\text{m}$  accuracy. Express in m, cm, mm and  $\mu\text{m}$  using decimals and also in m using scientific notation.

decimal: \_\_\_\_\_ m = \_\_\_\_\_ cm = \_\_\_\_\_ mm = \_\_\_\_\_  $\mu\text{m}$

scientific notation: \_\_\_\_\_ m

4. Measure dimensions of lab room (length, width, height) and the calculate the room volume in  $\text{m}^3$ . Express in meters. Aim for 1 cm = 0.01 m accuracy. An 8 m steel tape measure is available.

\_\_\_\_\_ m x \_\_\_\_\_ m x \_\_\_\_\_ m  $V_{\text{room}} =$  \_\_\_\_\_  $\text{m}^3$

5. Measure the diameter of each of 5 marbles with 20  $\mu\text{m}$  accuracy. Express in mm. (Hint: about 15.00 mm)

--	--	--	--	--

Calculate the average and standard deviation of your 5 measurements. You can use a computer or phone application or go to <http://yosemitefoothills.com/Calculator/> and use my on-line application.

average d: \_\_\_\_\_ mm +/- \_\_\_\_\_ mm

6. Calculate the average volume of your 5 marbles. Express your result as  $\text{mm}^3$ , and  $\text{cm}^3$ . (Hint: about  $1767 \text{ mm}^3 = 1.767 \text{ cm}^3$ )

$$V_{\text{average}} = \frac{4}{3} \pi r^3 = \frac{1}{6} \pi d^3 = \text{_____ mm}^3 = \text{_____ cm}^3$$

7. Weigh the 5 marbles one at a time with a digital scale (that I will provide) that has 0.01 gram (g) precision. (Hint: about 4.00 g)

--	--	--	--	--

8. Calculate the average and standard deviation of your 5 mass measurements. You can use a computer or phone application or go to <http://yosemitefoothills.com/Calculator/> and use my on-line application.

$$m_{\text{average}} = \text{_____ g} \pm \text{_____ g}$$

9. Calculate the density of a typical marble by dividing your average mass in grams by your average volume in  $\text{cm}^3$ . Express your result with units of  $\text{g/cm}^3$  and in  $\text{kg/m}^3$ . (Hint:  $1 \text{ g/cm}^3 = 1000 \text{ kg/m}^3$ )

$$\rho_{\text{average}} = \frac{m_{\text{average}}}{V_{\text{average}}} = \text{_____ g/cm}^3 = \text{_____ kg/m}^3$$

### Ways of counting

10. Direct: number of ABC blocks = \_\_\_\_\_ x \_\_\_\_\_ x \_\_\_\_\_ + \_\_\_\_\_ = \_\_\_\_\_

11. Estimate the number of marbles by mass = (total mass – jar mass)/your average marble mass.

$$\text{mass of marbles } m_{\text{marbles}} = m_{\text{jar with marbles}} - m_{\text{jar}} = \text{_____ g} \quad (\text{tare capability used on scale})$$

$$N \approx \frac{m_{\text{marbles}}}{m_{\text{average}}} = \text{_____}$$

12. Number of marbles by volume = (volume water displaced) / your average marble volume

$$\text{mass of water in filled jar } m_{\text{water in filled jar}} = m_{\text{jar full of water}} - m_{\text{jar}} = \text{_____ g}$$

$$\text{mass of water and marbles in jar } m_{\text{water + marbles}} = m_{\text{jar+marbles+water}} - m_{\text{jars}} = \text{_____ g}$$

$$m_{\text{displaced water}} = m_{\text{water in filled jar}} - (m_{\text{water+marbles}} - m_{\text{marbles}}) = \text{_____ g}$$

$$V_{\text{marbles}} = V_{\text{displaced water}} = \frac{m_{\text{displaced water}}}{\rho_{\text{water}}} = \frac{m_{\text{displaced water}}}{0.997 \text{ g/cm}^3} = \text{_____ cm}^3 \quad N \approx \frac{V_{\text{marbles}}}{V_{\text{average}}} = \text{_____}$$

$$\text{Note: } 1 \text{ mL} = 0.001 \text{ L} = 1 \text{ cm}^3 = 10^{-6} \text{ m}^3$$

13. Number of marbles by counting (shared effort of entire class):  $N = \text{_____}$

14. If the volume of an average grain of salt is about  $0.03 \text{ mm}^3$ , how big would a container of salt need to be that holds  $6.02 \times 10^{23}$  grains of salt? Hint: Multiply the volume of a grain by the number of grains. Express your result in  $\text{km}^3$  and if it were in the shape of a cube, give the length of a side of that cube in km.

$$V_{\text{salt cube}} = \text{_____ mm}^3 = \text{_____ km}^3 \quad l_{\text{side}} = \sqrt[3]{V_{\text{salt cube}}} = \text{_____ km}$$

15. How much mass would that amount have if the density of salt is  $\rho_{\text{salt}} = 2.16 \text{ g/cm}^3 = 2160 \text{ kg/m}^3$  ?  
Hint: Multiply the volume by this density and adjust units. Express your answer in kg.

$$m_{\text{salt cube}} = \rho_{\text{salt}} \cdot V_{\text{salt cube}} = \text{_____ kg}$$